001 09/29/93 BORIC ACID INSECTICIDE

PRODUCT NAME. BORIC ACID INSECTICIDE

MSDE #- BX2320

1. CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Product Name. BORIC ACID INSECTICIDE

Grades. Technical, SQ, NF

Chemical Formula H3BO3

Chemical Name/Synonyms. Boric Acid, Orthoboric Acid, Boracic Acid

Chemical Family Inorganic Borates
CAS Registry Number. 10043-35-3
TSCA Inventory Number 10043-35-3
EPA Posticide Rog. No. 1624-117

2. COMPOSITION/INFORMATION ON INGREDIENTS' OSHA HAZARDS

This product contains greater than 99 percent (%) Bone Acid (H3BO3). Bone Acid is hazardous under the OSHA Hazard Communication Standard based on animal chrome toxicity studies. Refer to Sections 3 and 11 for details on hazards.

3. HAZARD IDENTIFICATION EMERGENCY OVERVIEW:

Boric Acid is a white odorless, powdered substance that in not flammable, combustible, or explosive and it presents no unusual hazard if involved in a fire. Boric Acid presents little or no hazard (to humans) and has low acute oral and dermal toxicities. Care should be taken to minimize the amount of Boric Acid released to the environment to avoid ocological effects.

POTENTIAL ECOLOGICAL EFFECTS:

Large amounts of Boric Acid can be harmful to becon-sensitive plants and other ecological systems

POTENTIAL HEALTH EFFECTS:

Routes of Exposure. Inhalation is the most significant route of exposure in occupational and other settings. Dermal exposure is not usually a concern because Boric Acid is not absorbed through intact skin.

Inhalation: Occasional mild irritation effects to nose and throat may occur from inhalation of Bonc Acid dusts at levels greater than 10 mg/m3

Eye Contact: Borle Acid is non-instating to eyes in normal industrial use

Skin Contact: Boric Acid does not cause pritation to intact skin

Ingestion: Products containing Boric Acid are not intended for ingestion. Boric Acid has a relatively low acute toxicity. Small amounts (e.g. a teaspoonful) swallowed accidentally are not likely to cause effects, swallowing amounts larger than that may cause gestrointestinal symptoms

Cancer: Boric Acid did not cause cancer in long-term animal studies, and is not considered a carcinogen.

Reproductive: Long-term, high dose animal ingection studies have demonstrated reproductive effects in male animals. A human study of occupational exposure to botate dust showed no adverse effect to reproduction

Developmental: High dose animal ingostion studies have demonstrated developmental effects in fetuses of pregnant animals, including fetal weight loss.

Target Organs: No target organ has been identified in humans. High dose animal ingestion studies indicate the testes are the target organs in male animals.

Signs and Symptoms of Exposure: Symptoms of accidental over-exposure to Boric Acid have been associated with ingestion or by absorption through large areas of damaged skin. These may include nauses, vonuting, and diarrhes, with delayed effects of skin redness and peeling.

Refer to Section 11 for details on Toxicological Data.

4. FIRST AID MEASURES

Inhalation: No specific treatment is necessary since Bone Acid in not likely to be hazardous by inhalation. Prolonged exposure to dust levels in excess of regulatory limits should always be avoided.

Eye Contact: Use eye wash fountain or fresh water to cleanse eye. If irritation possists for more than 30 minutes, seek medical attention.

Skin Contact: No treatment necessary because non-irritating

Ingestion, Swallowing less than one teaspoon will cause no harm to healthy adults. If larger amounts are swallowed, give two glasses of water to drink and seek medical attention.

NOTE TO PHYSICIANS: Observation only is required for adult ingestion of less than 6 grams of Boric Acid. For ingestion in excess of 6 grams, maintain adequate kidney function and force fluids. Gastric lavage is recommended for symptomatic patients only. Hemodialysis should be reserved for massive acute migestion or patients with retail failure. Boric Acid analysis of urine or blood are only useful for documenting exposure and should not be used to ovaluate severity of poisoning or to guide treatment. (Further Information: Litovitz, T.L., Norman, S.A., Velici, J. C., Annual Report of the American Association of Poison Control Centers Data Collection System. Am. J. Emerg. Med. 1986; 4 427-458). 24 hour Medical consultation is available at (800) 228-5635 EXT. 144.

5. FIRE-FIGHTING MEASURES

General Hazard: None, because Boric Acid is not flammable, combustible or explosive. The product is itself a flame retardant.

Extinguishing Media: Any fite extinguishing media may be used on nearby fires
Flammability Classification (29 CFR 1910.1200): Non-flammable solid

Tion Commission (5) Carrier and Commission and

6. ACCIDENTAL RELEASE MEASURES

General: Borne Acid is a water-soluble white powder that may cause damage to trees or vegetation by root absorption. (Refer to Ecological Information Section 12 for specific information)

Land Spill: Vacuum, shovel or sweep up Boric Acid and place in containers for disposal in accordance with applicable local regulations. Avoid contamination of water bodies during clean up and disposal. No personal protective equipment is needed to clean up land spills.

Water Spill: Borio Acid will cause localized contamination of surrounding waters depending on the quantity dissolved in these waters. At high concentrations some damage to local vegetation, fish and other aquatic life may be expected. (Refer to Sections 12, 13 and 15 for additional information)

Bonc Acid in a non-hazardous wants when spilled or disposed of, as defined in the Resource Conservation and Recovery Act (RCRA) regulations (40 CFR 261) (Refer to Regulatory Information Section 15 for additional references and information regarding EPA and California regulations.)

7. HANDLING AND STORAGE

Storage Temperature: Ambient
Storage Pressure: Atmospheric
Special Sensitivity: Moisture (Caking)

General: No special handling precautions are required, but dry, indeed storage is recommended. To maintain package integrity and to minimize caking of the product, bags should be handled on a "first-in first-out" basis. Good housekeeping procedures should be followed to minimize dust generation and accumulation

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Engineering Controls: Use local exhaust ventilation to keep authorne concentrations of Bone Acid dust below permissible exposure levels

Personal Protection: Where authorne concentrations are expected to exceed exposure limits, NIOSH/MSHA certified respirators must be used. Eye goggles and gloves are not required for normal industrial exposures, but may be warranted if environment in excessively dusty.

Occupational Exposure Limits: Boric Acid is listed/regulated by OSHA, Cal OSHA and ACGIH an "Particulate Not Otherwise Classified" or "Nuisance Dust."

OSHA: PEL* 15 mg/m3 total dust and 5 mg/m3 respirable dust

ACGIH: TLV** 10 mg/m3 Cal OSHA: PBL* 10 mg/m3

*PEL = "Permissible Exposure Limit"

**TLV = "Threshold Limit Value"

9. PHYSICAL AND CHEMICAL PROPERTIES

Appearance: White, odorless, crystalline solid

Specific Gravity: 1 51

Vapor Pressure: Negligible @ 200C

Solubility in Water: 4.7% @200C; 27.58k @1000C

Melting Point: 170.90C (3400F) (Heated in closed space)
Viscosity: Not applicable Bolling Point Not applicable

Flash Point: None

PTI: 6.1 (0.1% solution), 5.1 (1.0% solution), 3.7 (4.7% solution)

@ 200C

Formula Weight: 61.84

10. STABILITY AND REACTIVITY

General: Boric Acid is a stable product, but when heated it loses water, first forming Meta boric Acid (HB02), and on further heating it in converted into Boric Oxide (B203).

Incompatible Materials and Conditions to Avoid:

Boric Acid reacts as a weak acid, which may cause corresion of base metals. Reaction with strong reducing agents such as inetal hydrodes or alkali metals will generate hydrogen gas which could create an explosive baserd.

Hazardous Decomposition: None

11. TOXICOLOGICAL INFORMATION

INGESTION (ACUTE ORAL TOXICITY): Low acute oral toxicity LD50 of Boric Acid in rats is 3500 to 4100 mg/kg of body weight.

SKIN (ACUTE DERMAL TOXICITY): Low acute dermal toxicity, LD50 of Bonc Acid in rabbits is greater than 2000 mg/kg of body weight Boric Acid is not absorbed through intact skin

PRIMARY SKIN IRRITATION INDEX: 0 (zero). Bonc Acid is non-corrosive

EYE: Draize test in rabbite produced mild eye tritiation effects. Fifty years of occupational exposure history reflect no indication of human eye injury from exposure to Boric Acid

INITALATION: Human epidemiological studies show no increase in pulmonary disease in occupational populations with chronic exposures to Boric Acid dust and Sodium Burate dust

CARCINOGENICITY: A Technical Report issued by the National Toxicology Program showed "no evidence of carcinogenicity" from a full 2-year bloassay on Boric Acid in mice at feed dozes of 2500 and 5000 PPM in the diet. No mutagenic activity was observed for Boric Acid in a recent battery of four short-term mutagenicity assays

REPRODUCTIVE/DEVELOPMENTAL TOXICITY: Animal studies indicate Borle Acid reduces or inhibits sperm production, causes testicular atrophy, and when given to pregnant animals during gestation may cause developmental changes. These feed studies were conducted under chronic exposure conditions leading to doses many times in excess of those that could occur through inhalation of dust in occupational settings.

REPRODUCTIVE TOXICITY (Fertility): Dictary Boric Acid levels of 6,700 PPM in chronic feeding studies in rats and dogs produced testicular atrophy, while dogs and rats receiving 2000 PPM did not develop testicular changes IWeir, Fisher, 1972. In chronic feeding studies of mice on thick containing 5000 PPM (550 mg/kg/d) Boric Acid, testicular atrophy was present, while nince fed 2500 PPM (275 mg/kg/d) Boric Acid showed no significant increase in testicular atrophy 2NTP, 1987. In another Boric Acid chronic study, in mice given 4500 PPM (636 mg/kg/d), degeneration of seminiferons tubules was present together with a reduction of germ cells, while at 1000 PPM (152 mg/kg/d) no effect was seen 3fail et al., 1991.

In a reproduction study on rats, 2000 PPM of dietary Bonc Acid had no adverse effect on lactation, litter size, weight and appearance 1 Weir, Fisher 1972. In a continuous breeding study in mice there was a reduction in fertility rates for males receiving 4500 PPM (636 mg/kg/d) Boric Acid but not for females receiving 4500 PPM Boric Acid 3Fail et al., 1991.

DEVELOPMENTAL TOXICITY: Butic Acid at dietary levels of 1000 PPM (78 mg/kg/d) administered to pregnant female rats throughout gestation caused a slight reduction in fetal weight, but was considered to be close to the NOAEL. Doses of 2000 PPM (163 mg/kg/d) and above caused fetal malformations and maternal toxicity. In mice the no effect level for fetal weight reduction and national toxicity was 1000 PPM (248 mg/kg/d) Boric Acid. Fetal weight loss was noted at dietary Boric Acid levels of 2000 PPM (452 mg/kg/d) and above. Malformations (agenesis or shortening of the thirteenth rib) were seen at 4000 PPM (103 mg/kg/d), 4ffeinted et al. 1992.

- 1 Weir, R.J. and Fisher, R.S., Toxicol Appl, Pharmacol., 23.351 364 (1972)
- National Toxicology Program (NTP) Technical Report Sories No TR324, NHIE Publication No 88-2580 (1987), PB88-213475/XAB
- 3 Fail et al , Fund Appl Toxicol 17, 225-239 (1991)
- 4 Heindel et al., Fund Appl. Toxicol 18, 266 277 (1992).

12. ECOLOGICAL INFORMATIONECOTOXICITY DATA:

Phytotoxicity: Although boron is on essential micronutrient for healthy growth of plants, it can be harmful to boron-sensitive plants in higher quantities. Plants and trees can be exposed by root absorption to toxic levels of boron in the form of water-soluble borate leached into nearby soil or waters. Care should be taken to minimize the amount of borate product released to the convironment.

Fish Toxicity: Boron naturally occurs in seawater at an average concentration of 5 mg B/liter. In laboratory studies the acute toxicity (96-lnr LC50) for under-yearling Colio salmon (Onchorbynchus kisutch) in seawater was determined as 40 mg B/L (added an acdum metaborate).

Boron concentrations in fresh surface waters are generally less than 1 mg B/L. Laboratory studies on the toxicity of freshwater fish were determined using early life (embryo-larval) stages in natural water and Boric Acid as a test substance. The results were

Rainbow trout (S. gaurdueri) 24-day LC50-150 0 mg B/L

36-day NOEC-LOEC-0 75-1 mg B/L

Goldfish (Carassius autatus) 7-day NOEC-LOEC -26 50 mg B/L

3-day LCSO 178 mg B/L

Invertebrate Toxicity: The acute toxicity (48-hom LC50) to Daphnida (Daphnia magna Straus) in natural water is reported to be 133 mg B/L (added as Boric Acid). Estimated chronic toxicity (21-day NOEC-LOEC) values of 6-13 mg B/L (added an Boric Acid) have also been reported.

ENVIRONMENTAL PATE DATA:

Persistence/Degradation: Boun is naturally occurring and ubiquitous in the environment. Boric Acid decomposes in the environment to natural borate.

Octanol/Water Partition Coefficient: log Pow. -0.7570 at 250C

Soil Mobility: The product is soluble in water and is leachable through normal soil

NOTE: Boron (B) is the element in Borne Acid which is used to characterize borate product ecological effects. To convert Boric Acid data to Boron (B), multiply by 0.1748

13. DISPOSAL CONSIDERATIONS

Disposal Guidance: Small quantities of Borte Acad can usually be disposed of at Municipal Landfills sites. No special disposal treatment is required, but refer to state and local regulations for applicable site-specific requirements. Tonnage quantities of product are not recommended to be sent to landfills. Such product should, if possible, be re-used for an appropriate application.

RCRA (40 CFR 261)) Boric Acid is not listed under any sections of the Federal Resource Conservation and Recovery Act (RCRA). California Hazardous Waste Designation: California identifies substances with acute LD50's less than 5000 mg/kg as "hazardous wastes". Boric Acid is therefore a "instantous waste" if spilled in California, and should be handled in accordance with applicable state regulations.

Refer to Section 15 for additional regulatory information

14. TRANSPORT INFORMATION

DOT Hazardous Material Classification:Boric Acid is not a U.S. Department of Transportation (DOT) Hazardous Material.

DOT Hazardous Substances Classification: Bonc Acid is not a DOT Hazardous Substance.

International Transportation: Borne Acid has no UN Number, and is not regulated under international rail, highway, water or arr transport regulations.

15. REGULATORY INFORMATION

TSCA No.: (10043-35-3) Borio Acid appears on the EPA TSCA inventory list.

PIFRA:
Boric Acid is registered with the EPA, in accordance with Section 3 of FIFRA as a pesticide product.
Refer to EPA approved product label for additional product Hazard and Precautionary information.

RCRA: Boric Acid in not listed as a hazardous wasto under any sections of the Resource Conservation and Recovery Act or regulations (40 CFR 201 et seg.).

Superfund: CERCLA/SARA Boric Acid is not listed under CERCLA (the Comprehensive Environmental Response Compensation and Liability Act) or its 1986 amendments, SARA, (the Superfund Amendments and Reauthorization Act), including substances listed under Section 313 of SARA, Toxic Chemicals, 42 USC 11023, 40 CFR 372 65, Section 302 of SARA, Extremely Hazardous Substances, 42 USC 13002, 40 CFR 355; or the CERCLA Hazardous Substances list, 42 USC 9604, 40 CFR 302

Safe Drinking Water Act: Boric Acid is not regulated under the SDWA, 42 USC 30Gg-1, 40 CFR 141 et seg. Consult state and local regulations for possible water quality advisories regarding boron

Clean Water Act (Federal Water Pollution Control Act): 33 USC 1251 et seq.

- (a) Boric Acid is not itself a discharge covered by any water quality criteria of Section 304 of the CWA, 33 USC 1314.
- (b) It is not on the Section 307 List of Priority Pollutants, 33 USC 1317, 40 CFR 129
- (c) It is not on the Section 311 List of Hazardous Substances, 33 USC 1321, 40 CFR 110.
- (d) It is not on the Section 311 List of Hazardous Substances, 33 USC 1321, 40 CFR 116.

OSHA/Cai OSHA: This MSDS document meets the requirements of both OSHA (29 CFR 1910 1200) and Cal OSHA (Title 8 CCR 5194(g)) hazard communication standards.

Refer to Section 8 for regulatory exposure limits.

IARC: The International Agency for Research on Cancer (of the World Health Organization) does not list or categorize Bone Acid as a caroinogen.

NTP Annual Report ou Carcinogens: Boric Acid is not listed

OSIIA Carcinogen: Bone Acid is not listed.

California Proposition 65: Boxic Acid in not listed on any Proposition 65 hats of carcinogens or reproductive toxicants

16. OTHER INFORMATION

Product Label Text Hezard Information: Refer to EPA approved product label for additional product Hazard and Precontingary information.

National Fire Protection Association (NFPA) Classification:

Health Flammability Reactivity

Hazardous Materials Information Systems (HKIS):

Red [.]	(Flammability)	0
Yellow:	(Reactivity)	0
Blue:	(Acute Health)	1*
* Chroni-	c Effects	

0

0

a

VAN WATERS & <u>ROGERS INC.</u> ("VW&R") EXPRESSLY DISCLAIMS ALL EXPRESS OR IMPLIED WARRANTIES OF MERCHANTABILITY AND FITNESS FOR A PARTICULAR PURPOSE, WITH RESPECT TO THE PRODUCT OR INFORMATION PROVIDED HEREIN, AND SHALL UNDER NO CIRCUMSTANCES BE LIABLE FOR INCIDENTAL OR CONSEQUENTIAL DAMAGES

ALL INFORMATION APPEARING HEREIN IS BASED UPON DATA OBTAINED FROM THE MANUFACTURER AND/OR RECOGNIZED TECHNICAL SOURCES. WHILE THE INFORMATION IS BELIEVED TO BE ACCURATE, VW&R MAKES NO REPRESENTATIONS AS TO HS ACCURACY OR SUFFICIENCY. CONDITIONS OF USE ARE BEYOND VW&R'S CONTROL AND THEKEFORE USERS ARE RESPONSIBLE TO VBRIFY THIS DATA UNDER THEIK OWN OPERATING CONDITIONS TO DETERMINE WHETHER THE PRODUCT IS SUITABLE FOR THEIR PARTICULAR PURPOSES AND THEY ASSUME ALL RISKS OF THEIR USE, HANDLING, AND DISPOSAL OF THE PRODUCT, OR FROM THE PUBLICATION OR USE OF, OR RELIANCE UPON, INFORMATION CONTAINED HEREIN THIS INFORMATION RELATES ONLY TO THE PRODUCT DESIGNATED HEREIN, AND DOES NOT RELATE TO ITS USE IN COMBINATION WITH ANY OTHER MATERIAL OR IN ANY OTHER PROCESS.

END OF MSDS